# Chemical Constituents of Euphorbia marschalliana Boiss.

Amir Reza Jassbi<sup>a,b,\*</sup>, Simin Zamanizadehnajari<sup>b</sup>, and Satoshi Tahara<sup>b</sup>

- <sup>a</sup> Department of Phytochemistry, Medicinal Plants Research Institute, Shahid Beheshti University, Evin, Tehran, Iran. Fax: +98-21-2418679. E-mail: a-jassbi@cc.sbu.ac.ir
- b Laboratory of Ecological Chemistry, Division of Applied Bioscience, Graduate School of Agriculture, Hokkaido University, Kita-ku, Sapporo 060-8589, Japan
- \* Author for correspondence and reprint requests
- Z. Naturforsch. 59 c, 15-18 (2004); received April 28/June 23, 2003

Acetone extract of aerial parts of *Euphorbia marschalliana* collected from Iran has been subjected to different chromatography techniques for fractionation and purification. The stereo-structures of the myrsinol esters 15-*O*-acetyl-3-*O*-propionyl-5-*O*-butanoyl-7-*O*-nicotinoylmyrsinol (1) and 15-*O*-acetyl-3,5-*O*-dibutanoyl-7-*O*-nicotinoylmyrsinol (2) have been probed using ROESY spectroscopy and modified for the stereochemistry at C-6, C-12 and C-13.  $\beta$ -Sitosterol (3), 29-norcycloart-5-ene (4), 5,8-lanostadiene-3 $\beta$ -ol (5), 3 $\beta$ ,24( $\beta$ ),25-trihydroxycycloartane (6), 3 $\beta$ ,24( $\beta$ ),25-trihydroxycycloartane (7) and 24-methylenecycloartan-3 $\beta$ -ol (8) were identified for the first time in this plant.

Key words: Euphorbia marschalliana Boiss., Myrsinol Ester Diterpenoids, Cycloartane Triterpenoids

#### Introduction

The irritant ingenol and non-irritant myrsinol diterpene esters have been separated from *Euphorbia myrsinites* (Rentzea *et al.*, 1982; Rentzea and Hecker, 1982; Öksüz *et al.*, 1995).

Recently a new diterpenoid ester was isolated from E. decipiens, which showed inhibitory activity against prolyl endopeptidase, a serine protease, and analgesic activity (Ahmad et al., 2002). Investigation of the constituents of E. myrsinites of Turkey yielded four myrsinol esters with moderate anti-HIV-1 reverse transcriptase (RT) inhibitory activity (Öksüz et al., 1995). The above biological activities and the inhibitory activity of cheiradone, a myrsinol-type diterpenoid isolated from E. cheiradenia, against  $\alpha$ -glucosidase (Abbas et al., 2000) prompted us to study the constituents of Euphorbia marschalliana of Iran. Determination of the stereochemistry of the myrsinane-type diterpenoids was other objective of this research. These types of research seem to be essential since in different articles opposite stereochemistry is reported for myrsinol-type diterpenoids (Rentzea et al., 1982; Rentzea and Hecker, 1982; Öksüz et al., 1995). Recently our group determined the absolute configuration of decipinone, a myrsinanetype diterpenoid, by NMR spectroscopy (Jassbi et al., 2002).

This article reports the isolation of two myrsinol esters, 15-*O*-acetyl-3-*O*-propionyl-5-*O*-butanoyl-7-

O-nicotinoylmyrsinol (1) and 15-O-acetyl-3,5-O-dibutanoyl-7-O-nicotinoylmyrsinol (2), and the determination of their relative configuration.  $\beta$ -sitosterol (3), 29-norcycloart-5-ene (4), 5,8-lanostadiene-3 $\beta$ -ol (5), 3 $\beta$ ,24(S),25-trihydroxycycloartane (6), 3 $\beta$ ,24(R),25-trihydroxycycloartane (7) and 24-methylenecycloartan-3 $\beta$ -ol (8) were identified for the first time in the acetone extract of E. marschalliana.

# **Results and Discussion**

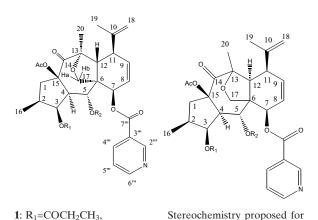
The acetone extract of the aerial parts of the plant *E. marschalliana* was subjected to column chromatography on silica gel (70–230 mesh).

Two myrsinol esters (1, 2), five triterpenes and  $\beta$ -sitosterol were separated from semi-polar fractions. The planar structures of 1 and 2 were determined by EI and FD mass and <sup>1</sup>H and <sup>13</sup>C NMR spectra and confirmed by comparison with those reported for these compounds previously (Öksüz et al., 1995). The stereochemistry of myrsinol esters isolated from E. myrsinites were determined by X-ray crystallography but the orientation of H-5 was determined as alpha despite the large coupling constant between H-4 and H-5 (J =11.5 Hz) (Rentzea et al., 1982). Table I shows the <sup>1</sup>H NMR and ROESY spectral data for compound 1. The carbon skeletons of 1 and 2 (Fig. 1) were suggested by Öksüz et al. (1995). Besides <sup>1</sup>H NMR and <sup>13</sup>C NMR, selective INEPT spectroscopy ap-

Table I. 1H NMR and ROESY spectral data for compound 1.

Position	<sup>1</sup> H NMR (in CDCl <sub>3</sub> )	NOE correlations using ROESY spectrum
1α	3.50 dd (9.6, 14.8)	
$1\beta$	1.4 m	
2 3	2.2 m	
	5.21 t (3.7)	H-2', H-4, H-16
4	2.29 dd (3.5, 11.1)	$H-3, H_a-17$
5	6.07 d (11.1)	H-7, H-12
7	5.11 d (6.4)	$H-5$ , $H-8$ , $H-9$ , $H_b-17$ , $H_a-17$
8	6.21 ddd (1.5, 6.4, 9.6)	H-7, H-11, H <sub>b</sub> -17, H-18
9	5.91 dd (5.7, 9.6)	H-7, H-11, H <sub>b</sub> -17
11	3.36 brs	H-9, H-18, H-20,
12	3.26 d (4.6)	H-5, H-18, H-19,
16	0.88 d (6.9)	
17 <sub>a</sub>	3.97 d (9.6)	H-4, H-7, H <sub>b</sub> -17
$17_{\rm b}$	3.60 d (9.6)	
18 <sub>a</sub>	4.80 brs	H-11, H-19, H-20
18 <sub>b</sub>	4.78 brs	
19	1.73 brs	
20	1.56 s	
OAc	2.19 s	
Propionyl		
2'	2.22 q (7.4)	
3'	1.03 t (7.4)	
Butanoyl	,	
2"	2.14 t	
3"	1.43 m	
4"	0.83 t (7.4)	
Nicotinoyl	,	
2""	9.14 s	H-5
4‴	8.21 ddd (1.7, 1.7, 8.0)	H-5
5'''	7.37 dd (5.0, 8.0)	
6'''	8.74 dd (1.7, 5.0)	

plied to determine the relative positions of the esters. The small coupling constant between H-12 and H-11 (J = 3.5 Hz) may be the reason of sug-



R<sub>2</sub>=COCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> compounds 1 and 2 by Öksüz 2: R<sub>1</sub>=R<sub>2</sub>=COCH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> et al. (1995).

Stereochemistry proposed for

Fig. 1. Structure of myrsinane diterpenoids from E. marschalliana (left) and E. myrsinites (right).

gestion of alpha orientation for H-12 for compounds **1** and **2** (Fig. 1, right) (Öksüz *et al.*, 1995). In this article we determined the structures of 1 and 2 using <sup>1</sup>H NMR, <sup>13</sup>C NMR, HMQC and HMBC spectral data (Fig. 1, left). In order to investigate the stereochemistry of 1 isolated from E. marschalliana we additionally used 2D ROESY spectroscopy. In the ROESY spectrum of 1 the strong cross peaks between H-12 and H-5, H-5 and H-2", H-5 and H-4" in the nicotinoyl moiety and between H-12 and H-19 suggested that these protons have the same orientation in the molecule. On the other hand the NOE correlations between H<sub>a</sub>-17 and H-4 and H-7, H-8 and H-9 with H<sub>b</sub>-17 and between H-20 and H-11 suggested that these protons are located on the other face of the molecule (Fig. 2).

Isolation of cycloartane triterpenes and myrsinol esters from both E. marschalliana of Iran and E. myrsinites of Turkey (Öksüz et al., 1995) indicates the close chemical relationship between these two plants. Identification of compounds 3-

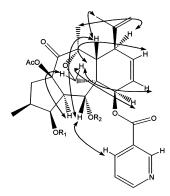


Fig. 2. NOE correlations using ROESY spectral data for compound 1.

**8** was confirmed by their mass spectra, <sup>1</sup>H NMR and in some cases <sup>13</sup>C NMR spectra.

## **Experimental**

General

 $^{1}$ H and  $^{13}$ C NMR spectra (broad band and DEPT experiments) were measured on a JEOL, JNM-EX270 instrument.  $^{1}$ H NMR, HMQC, HMBC and ROESY spectra of compound 1 were measured on a 500 MHz Bruker AMX500. MS spectra were recorded on a JEOL JMS-SX102A spectrometer. Analytical TLC experiments were performed on Merck silica gel 60  $F_{254}$  and DIOL  $F_{254}$  HPTLC pre-coated glass plates. Column chromatographies were performed, on silica gel (63–210 μm and 40–60 μm) and LiChroprep Diol (40–60 μm).

### Plant material

The aerial parts of the plant *Euphorbia marschalliana* Boiss. were collected from Taleghan area near Karaj, Iran in April 2000. The plant was identified by Mr. Bahram Zehzad at the Department of Biological Sciences, Shahid Beheshti University, Evin, Tehran, Iran. The voucher specimens were deposited in the herbarium of the same department (herbarium no. 2000–913, 98–104).

Extraction of the plant material and purification of the compounds

After grinding, the shade dried plant material (2 kg) was extracted with acetone for one week. The acetone extract was concentrated under reduced pressure and partitioned between water/

chloroform and water/*n*-butanol. The chloroform layer (54.5 g) was subjected to column chromatography using silica gel  $(63-210 \,\mu\text{m}, 550 \,\text{g})$ . The column was eluted with n-hexane with increasing polarity up to ethyl acetate followed by methanol. Compounds 1 and 2 were purified from fractions 12 and 13 eluted with about 30% ethyl acetate in hexane. 1.4 g of the above fractions were subjected to column chromatography using diol stationary phase (80 g), the column was eluted with hexane followed by increasing the polarity using chloroform and 10% methanol in chloroform. The fractions were eluted with 70% chloroform in hexane were subjected to preparative silica gel TLC using ethyl acetate/toluene (3.5:6.5) and chloroform/acetone/acetic acid (96:3:1 v/v/v) to afford 1 (6 mg) and 2 (41 mg), respectively. The fractions were eluted (from the first column) with hexane/ethyl acetate (predominant in hexane) were mixed (2.08 g) and load (0.70 g) on a column containing 8% AgNO3 on silica gel  $(40-60 \,\mu\text{m}, 28 \,\text{g})$ . The column was eluted with hexane/chloroform (1:1) with increasing the polarity from chloroform to 10% acetone in chloroform. Compounds 4 and 5 (4 mg) were semi-purified as a mixture. Fraction 11 (2.24 g) eluted with 15% ethyl acetate in hexane was subjected to column chromatography  $(40-60 \, \mu \text{m}, 66 \, \text{g})$  using hexane with increasing the polarity up to chloroform and 10% acetone in chloroform. Repeated preparative silica gel TLC of chloroform rich fractions using 5% acetone in chloroform and 20% ethyl acetate in chloroform yielded compounds 3 (3.5 mg) and 8 (5.2 mg). Compound 7 (534 mg) was purified from more polar fractions prominent in ethyl acetate using column chromatography over flash silica gel using 80% chloroform in hexane with increasing the polarity to pure acetone.

Spectral data for the identified compounds

Compound 1: EIMS: m/z (rel. int., %) = 638 (3) [M+1]<sup>+</sup> (C<sub>35</sub>H<sub>43</sub>NO<sub>10</sub>), 609 (16), 578 (16), 567 (22), 549 (14), 521 (44), 166 (83), 124 (100), 106 (26). – <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) see the Table I. – <sup>13</sup>C NMR (67.5 MHz, CDCl<sub>3</sub>):  $\delta$  = 42.2 (C-1), 36.8 (C-2), 77.1 (C-3), 52.9 (C-4), 66.3 (C-5), 55.4 (C-6), 66.0 (C-7), 121.8 (C-8), 134.8 (C-9), 146.0 (C-10), 41.8 (C-11), 42.9 (C-12), 90.0 (C-13), 203.1 (C-14), 90.4 (C-15), 14.4 (C-16), 69.9 (C-17), 113.3 (C-18), 21.1 (C-19), 20.8 (C-20), CH<sub>3</sub>CO: 20.0, CH<sub>3</sub>CO: 169.7, CH<sub>3</sub>CH<sub>2</sub>CO: 173.4 (C-1'), 27.8 (C-2'), 8.8

(C-3'), CH<sub>3</sub>CH<sub>2</sub> CH<sub>2</sub>CO: 170.8 (C-1"), 35.8 (C-2"), 17.9 (C-3"), 13.6 (C-4"), C<sub>6</sub>H<sub>4</sub>NO: 150.5 (C-2"'), 126.3 (C-3"'), 136.8 (C-4"'), 122.9 (C-5"'), 152.8 (C-6"'), 164.2 (C-7"').

Compound 2: EIMS: m/z (rel. int., %) = 652 (2)  $[M+1]^+$  (C<sub>36</sub>H<sub>45</sub>NO<sub>10</sub>), 581 (9), 535 (18), 405 (12), 282 (14), 264 (14), 124 (100). – FDMS m/z = 651 $[M]^+$ , 623  $[M^+$ -CO]. – <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>):  $\delta = 3.51$  (H-1 $\alpha$ , dd, J = 9.5, 14.8 Hz), 5.22 (H-3, t, J = 3.8 Hz), 2.30 (H-4, dd, J = 3.6, 11.2 Hz),6.09 (H-5, d, J = 11.2 Hz), 5.13 (H-7, d, J = 6.4 Hz),6.23 (H-8, ddd, J = 2.0, 6.4, 9.8 Hz), 5.93 (H-9, dd, J = 5.8, 9.8 Hz), 3.38 (H-11, brs), 3.27 (H-12, d, J =3.5 Hz), 0.83–0.92 (m, 9H, H-16, H-4', H-4"), 3.99  $(H_a-17, d, J = 9.6 Hz), 3.62 (H_b-17, d, J = 9.6 Hz),$ 4.83 (H<sub>b</sub>-18, brs), 4.80 (H<sub>b</sub>-18, brs), 1.76 (H-19, s), 1.58 (H-20, s), CH<sub>3</sub>CO: 2.21 (s), CH<sub>3</sub>CH<sub>2</sub>CH<sub>2</sub>CO: 2.13-2.19 (H-2', H-2", m), 1.38-1.54 (H-3', H-3", m),  $C_6H_5NO$ : 9.16 (H-2", s), 8.23 (H-4", dt, J =2.0, 6.2 Hz), 7.39 (H-5", dd, J = 5.0, 8.0 Hz), 8.75 (H-6''', brs).

Compound 3: EIMS: m/z (rel. int., %) = 414 (100) [M]<sup>+</sup> (C<sub>29</sub>H<sub>50</sub>O), 399 (17), 396 (25), 381 (12), 329 (17), 303 (15). – <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>):  $\delta$  = 5.35 (brd, J = 5.4 Hz), 3.52 (m, 1H), 1.00 (s, 3H), 0.92 (d, J = 6.6 Hz, 3H), 0.83 (d, J = 6.0 Hz, 6H), 0.81 (t, J = 6.0 Hz, 3H), 0.68 (s, 3H). – <sup>13</sup>C NMR (67.5 MHz, CDCl<sub>3</sub>):  $\delta$  = 140.0, 121.6, 71.8, 56.8, 56.1, 50.2, 45.9, 42.4, 42.4, 39.8, 37.3, 36.6, 36.2, 34.0, 31.9, 31.9, 31.8, 29.2, 28.3, 26.2, 24.4, 23.2, 21.1, 19.9, 19.5, 19.1, 18.9, 12.1, 11.9.

Compounds **4**, **5**: FDMS: m/z (rel. int., %) = 426 [M]<sup>+</sup>. - <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>):  $\delta$  = 5.09 (t, J = 6.7 Hz), 3.25 (ddd, J = 4.3, 11.4, 11.4 Hz), 1.68 (s), 1.60 (s), 0.96 (s), 0.89 (s), 0.80 (s), 0.55 (d, J = 4.1 Hz), 0.32 (d, J = 4.1 Hz). - <sup>13</sup>C NMR (67.5 MHz, CDCl<sub>3</sub>):  $\delta$  = 134.4, 134.0, 133.5, 130.9, 125.3, 125.2, 79.0, 78.8, 52.3, 50.9, 50.4, 50.0, 49.6, 48.8,

47.9, 47.1, 36.3, 36.2, 35.8, 35.6, 35.4, 35.2, 32.8, 31.9, 30.9, 30.4, 29.9, 29.8, 29.7, 28.1, 27.9, 26.5, 26.1, 25.7, 25.4, 24.9, 24.7, 24.5, 21.5, 21.1, 20.1, 20.0, 19.3, 18.9, 18.2, 18.0, 17.7, 15.7, 15.6, 15.3, 15.4, 14.0.

Compounds **6**, **7**: FDMS: m/z (rel. int., %) = 460 [M]<sup>+</sup>. - <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>):  $\delta$  = 3.30 (m), 1.21 (s), 1.16 (s), 0.96 (s, 6H), 0.89 (s), 0.80 (s), 0.33 (d, J = 4.1 Hz), 0.55 (d, J = 4.1 Hz). - <sup>13</sup>C NMR (67.5 MHz, CDCl<sub>3</sub>):  $\delta$  = 79.6, 78.8, 78.7, 73.2, 73.1, 52.5, 52.3, 48.5, 48.0, 47.1, 45.4, 45.3, 40.5, 36.4, 35.9, 35.6, 33.6, 33.2, 32.9, 32.0, 30.4, 29.9, 28.8, 28.5, 28.3, 28.2, 26.7, 26.6, 26.5, 26.1, 26.0, 25.5, 23.3, 23.2, 21.2, 20.0, 19.4, 18.5, 18.2, 18.1, 14.1.

Compound **8**: EIMS: m/z (rel. int., %) = 440 (69) [M]<sup>+</sup> (C<sub>31</sub>H<sub>52</sub>O), 426 (30), 425 (53), 422 (100), 407 (73), 379 (31), 300 (58). – <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.71 (brs), 4.66 (brs), 3.28 (dd, J = 4.3, 10.7 Hz), 2.26 (m), 1.02 (d, J = 6.7 Hz), 1.03 (d, J = 6.7 Hz), 0.96 (s, 6H), 0.90 (s), 0.89 (d, J = 6.4 Hz), 0.81 (s), 0.32 (d, J = 4.1 Hz), 0.55 (d, J = 4.1 Hz). – <sup>13</sup>C NMR (67.5 MHz, CDCl<sub>3</sub>):  $\delta$  = 105.0, 78.8, 52.2, 48.8, 47.9, 47.1, 45.3, 40.4, 36.1, 35.5, 35.0, 35.8, 32.9, 31.9, 31.3, 30.4, 29.8, 29.6, 28.1, 26.4, 26.1, 26.0, 25.4, 21.9, 21.8, 21.1, 20.0, 19.3, 18.3, 18.0, 13.9.

### Acknowledgements

One of us (A. R. J.) is thankful to the "Japan Society for the Promotion of Science" for granting a post-doctoral fellowship. Our thanks are expressed to Dr. Eri Fukushi and Mr. Kenji Watanabe for their skilful measuring the NMR and MS spectra and Mr. Bahram Zehzad at the Department of Biological Sciences, Shahid Beheshti University, Evin, Tehran, Iran for identification of the plant material.

Abbas M., Jassbi A. R., Zahid M., Ali Z., Alam N., Akhtar F., Choudhari M. I., and Ahmad V. U. (2000), Three new diterpenoids from *Euphorbia cheiradenia*. Helv. Chim. Acta **83**, 2751–2755.

Ahmad V. U., Hussain H., Hussain J., Jassbi A. R., Bukhari I. A., Yasin A., Choudhary M. I., and Dar A. (2002), New bioactive diterpenoid from *Euphorbia decipiens*. Z. Naturforsch. **57b**, 1066–1071.

Jassbi A. R., Fukushi Y., and Tahara S. (2002), Determination of absolute configuration of decipinone, a diterpenoid ester with a myrsinane-type carbon skeleton, by NMR spectroscopy. Helv. Chim. Acta 85, 1706–1713.

Öksüz S., Gürek F., Gil R. R., Pengsuparp T., Pezzuto J. M., and Cordell G. A. (1995), Four diterpene esters from *Euphorbia myrsinites*. Phytochemistry **38**, 1457–1462.

Rentzea M. and Hecker E. (1982), α-Ketol-Umlagerung von Myrsinol zum Iso-myrsinol und mögliche Biogenese des Myrsinangerüstes. Tetrahedron Lett. 23, 1785–1788.

Rentzea M., Hecker E., and Lotter H. (1982), Neue tetrazyklische, polyfunktionelle Diterpenoide aus *Eu-phorbia myrsinites* L. Röntgenstrukturanalyse und Stereochemie des 14-Desoxo-14β-hydroxymyrsinols. Tetrahedron Lett. **23**, 1781–1784.